umdberg

Wiki Pages & Files

VIEW EDIT

Energy in photosynthesis - a toy model

| last edited by | Joe Redish 2 years ago | | Pa | ige history |
|----------------|------------------------|--|----|-------------|
| 6.4.P15 | | | | |
| Prerequisite | : | | | |

Energy in photosynthesis

Photosynthesis is the process by which plants effectively capture energy from the sun and store it for later use by organisms. It's the fundamental process that enables the entire food chain. Actual photosynthesis is a highly complex process involving many steps and many different chemicals. But the idea of what is happening to the overall energy balance can be seen even in simpler reactions.

Let's consider a toy model that shows how an input of energy (for example from the sun) could be used to combine water and carbon dioxide to create an organic molecule (formaldehyde) and oxygen and effectively "store" that energy for use later. The figure below shows a the simplest possible such process. The potential energy (binding deficit) of each bond is shown on the bonds.



A. First check the stoichiometry. Are the number of each kind of atoms balanced?

B. After you have balanced your equation, calculate the amount of bond energy deficit on each side of the equation and identify which side needs to have energy added to it to make the equation balance energetically and how much. What does it mean to say "you have to add energy to one side of the equation to make the energy balance?"

C. A particular direction of the reaction is labeled as exothermic. Is it correct? Explain how you know using some schematic visual representation of the energies (potential energy graphs or energy bars).

D. Discuss where the energy is "stored". This is not a question with a clear answer, since the energy relations in the reaction are a result of all the different bonds in all the different molecules. But discuss the two questions, "When the reaction goes in the exothermic direction and energy becomes available for use, where does the energy come from?" and "Where is it useful to think of the energy as being stored and why?" If these two answers are different, reconcile why it is OK for them to be different.

Follow ons:

<

- Energy in photosynthesis light and dark reactions
- Energy in photosynthesis photon absorption and efficiency

Joe Redish and Todd Cooke 11/21/14

Based on Photosynthesis, by E. Rabinowich and Govindjee, Chapter 5.

Comments (0)

You don't have permission to comment on this page.

Search this workspace

To join this workspace, request access.

| Already | have | an | account? | Log | in! | |
|---------|------|----|----------|-----|-----|--|
|---------|------|----|----------|-----|-----|--|

| Navigato | r | | | | |
|----------|-----------|----------|----------|---------|---|
| Open | 131-F13 (| Course N | laterial | | |
| | | | | | ^ |
| | | | | | |
| | | | | | |
| | | | | | |
| | | | | | |
| | | | | | |
| | | | | | |
| | | | | | ~ |
| Pages | Files | | | options | |

SideBar

BERG Home Who We Are Upcoming Meetings NEXUS Calendar NEXUS Development Page Phys 131-132 TA & LA Info Page Thermo Pages BERG Camera Signup Main Data Page (secure) Instructor Reflections

PERG Home ChERG Home

Recent Activity

Spring 2017 Phys 132 Training Reminders edited by Kim Moore

Spring 2017 Physics 132 Reminder from... edited by Kim Moore

Spring 2017 Physics 132 Reminder from... created by Kim Moore

Spring 2017 Phys 131 Training Reminders edited by Kim Moore

Spring 2017 Physics 131 Reminder from... edited by Kim Moore

Spring 2017 Physics 131 Reminder from... created by Kim Moore

Spring 2017 Physics 131 Reminder from... edited by Kim Moore

More activity...

Printable version

PBworks / Help Terms of use / Privacy policy

About this workspace ivacy policy Contact the owner / RSS feed / This workspace is public log in help